AP-42 Section Reference

A COMPREHENSIVE EMISSION INVENTORY REPORT AS REQUIRED UNDER THE AIR TOXICS "HOT SPOTS" INFORMATION AND ASSESSMENT ACT OF 1987

PREPARED FOR

CALMAT CO.
FRESNO NO. II FACILITY
FRESNO, CALIFORNIA

SEPTEMBER 14, 1990

SUBMITTED TO

FRESNO COUNTY AIR POLLUTION CONTROL DISTRICT 1221 FULTON MALL FRESNO, CALIFORNIA 93721

PREPARED BY

ENGINEERING-SCIENCE, INC. 75 North Fair Oaks Avenue P.O. Box 7107 Pasadena, California 91109

FRESNO

INDUSTRIAL ASPHALT PLANT

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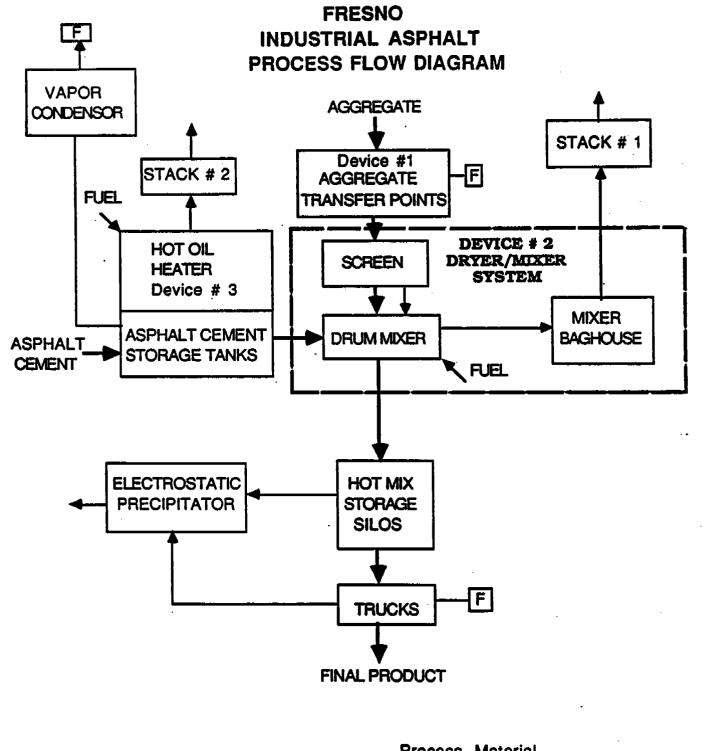
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ZING SJRAWOLIA LAMOITQO: RYNESJ	*CONTROL EQPT CODES* OVERALL FULLY HOURLY MAX EMISSIONS PRIMARY SECONDARY CONTROL EFFIX) PART (LBS/HOUR)	_				
	0.0.0 1 1 0.0.0 1 N. I.					
NAME A NOTE	a Assiam date 5-25-90	MO8/49031;				

AIR TOXICS EMISSION DATA SYSTEM REVIEW AND UPDATE REPORT PROCESS AND EMITTENTS DATA (ADDITIONAL EMITTENTS)				
OFFICE USE ONLY GO: 10 FACID: 10	EMITTENT DATA EMISSIONS EMITTENT ID EST ACTUAL EMISSIONS ANNUAL AVERAGE			
COS.	77 8:2:5:05: 1 2 N.D. C N.D.			
ALOWABLE EMIS LBS/YR-IOPTIONAL	*CONTROL EQPT CODES* OVERALL FULL/ PRIMARY SECONDARY CONTROL EFF(%) PART (LBS/HOUR) O O O O O O O O O O O O O O O O O O O			
ACTIONS: CODE:	EMITTENT ID J EST ACTUAL EMISSIONS ANNUAL AVERAGE EMISSIONS (LBS/YR)			
ALLOWABLE BIAS ESSATE OPTIONAL	*CONTROL EQPT CODES* OVERALL FULL/ PRIMARY SECONDARY CONTROL EFF(%) PART C C C C C C C C C C C C C C C C C C C			
ACTION CODE	EMITTENT ID EST ACTUAL EMISSIONS ANNUAL AVERAGE EMISSIONS (LSS/YR) C [V 40.4/7 2 N.D., D.			
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ACTION COSE	EMITTENT ID EST ACTUAL EMISSIONS ANNUAL AVERAGE EMISSIONS (LBS/YR) 744014391 2			
ALLOWASE EMIS ESTATE OPTIONALI THE LITTERS	PRIMARY SECONDARY CONTROL EFFIX) C C C C C C C C C C C C C	<u> </u>		
ACTION CODE		1 .		
LES/YR-IOPTIONALI THE HITCH	*CONTROL EQPT CODES* OVERALL PRIMARY SECONDARY CONTROL EFFIX) C C C C C C C C C C C C C C C C C C C			

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	7440666 2 1 2.79 E-04 6.97	
ALEOWABLE EMIS LBS/YR IOPTIONAL	+CONTROL EQPT CODES+ OVERALL FULL/ HOURLY MAX EMISSIONS PRIMARY SECONDARY CONTROL EFF(%) PART (LBS/HOUR)	_
	000 000125	_
ACTION CODE	EMITTENT ID EST ACTUAL EMISSIONS ANNUAL AVERAGE EMISSIONS (LBS/YR)	
ALLOWABLE EMIS	-CONTROL EQPT CODES- OVERALL FULLY HOURLY MAX EMISSIONS	
LESOYR (OPTIONAL)	PRIMARY SECONDARY CONTROL EFFINI PART (LBS/HOUR)	_
ACTION		
CODE	THE TIPLE OF THE PROPERTY OF T	
ALIGWARLE EMIS LESZYR TOPTIONALI	+CONTROL EQPT CODES+ OVERALL FULL/ HOURLY MAX EMISSIONS PRIMARY SECONDARY CONTROL EFFISH PART (LBS/HOUR)	
TETTE I	000 1 000 1 000 100 1009 E-04.	<u>.</u>]
FACTION:	EMITTENT ID EST ACTUAL EMISSIONS ANNUAL AVERAGE METH FACTOR (LBS/UNIT) EMISSIONS (LBS/YR)	
	743.9965. 2 12.50 E.06 1 2.0626	
ALLOWABLE EMIS ESSAYR (OPTIONAL)	PRIMARY SECONDARY CONTROL EFFIX) PART LESSHOURI	- ; │
	0.0.0 1.1 0.1010 1.1.1.3. Eno.5.	1
ACTION	EMITTENT ID V EST ACTUAL EMISSIONS ANNUAL AVERAGE EMISSICHS (LBS/YR)	
ALIDWARLE EMIS	•CONTROL EQPT CODES• OVERALL FULL/ HOURLY MAX EMISSIONS	<u> </u>
LBS/YR#IDPTIONAL)	PRIMARY SECONDARY CONTROL EFFIXI PART (LBS/HOUR) 1.00 C C C C C C C C C C C C C C C C C C	
NAME ANTOIN	1 Assion DITE 5-25-90	PR08/8903

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1,	YEAR AIR TO	OXICS EMISSION DATA SYSTEM REVIEW AND UPDATE REPORT PROCESS AND EMITTENTS DATA (ADDITIONAL EMITTENTS)	PRO			
1'	OFFICE USE ONLY CO / CO FACID YOS	DEVICE ID 3	3106 3			
j		EMITTENT DATA EMISSIONS				
	ACTION CODE	EMITTENT ID EST ACTUAL EMISSIONS ANNUAL AVERAGE EMISSIONS (LBS/YR)				
\$		7440020: 1 2 N.D.: 1 N.D.: .				
;	ALLOWABLE EVES LBS/YR IOPTIONAL	+CONTROL EQFT CODES+ OVERALL FULL/ HOURLY MAX EMISSIONS PRIMARY SECONDARY CONTROL EFF(%) PART (LBS/HOUR)	5			
-	MATTER	000 1 1 0 0 0 1 1 N.D.				
	ACTION	EMITTENT ID EST ACTUAL EMISSIONS ANNUAL AVERAGE				
: :	CODE:	7782492: 2 D.D. EMISSIONS (LBS/YR)				
	ALEOWASIE EMS	*CONTROL EQPT CODES* OVERALL FULL/ HOURLY MAX EMISSION: PRIMARY SECONDARY CONTROL EFF(%) PART (LBS/HOUR)	s			
		000 000				
:		•	-			
:	ACTION CODE	EMITTENT ID EST ACTUAL EMISSIONS ANNUAL AVERAGE EMISSIONS (LES/YR)				
		1165 2 N.D. N.D.				
:	ALLOWABLE ENS ESOMR (OPTIONAL)	CONTROL EQPT CODES* OVERALL FULL/ HOURLY MAX EMISSION: CONTROL EFF(%) PART (LES/HOUR)				
:	BELLE	DOOD TOOOD TOOOD	<u></u>			
,	ACTION CODE	EMITTENT ID EST ACTUAL EMISSIONS ANNUAL AVERAGE METH FACTOR (LBS/UNIT) EMISSIONS (LBS/YR)				
!						
	ALLOWASLE EMIS LESTYR HOPTIONAL	*CONTROL EOPT CODES* OVERALL FULL/ HOURLY MAX EMISSION PRIMARY SECONDARY CONTROL EFF(%) PART (LBS/HOUR)	s			
i						
 	AGTION	EMITTENT ID EST ACTUAL EMISSIONS ANNUAL AVERAGE				
	CODE	METH FACTOR (LBS/UNIT) EMISSIONS (LBS/YR)				
	ALLOWABLE EMIS	*CONTROL EQPT CODES* OVERALL FULLY HOURLY MAX EMISSION				
	LBS/YR IOPTIONAL	PRIMARY SECONDARY CONTROL EFFI%) PART (LBS/HOUR)				
į						
	NAME Anthina Assidium DATE 5-25-90					

EMISSION
YEAR
1989
1022

AIR TOXICS EMISSION DATA SYSTEM REVIEW & UPDATE REPORT SUPPLEMENTAL PROCESS PARAMETER FORM SUBSTANCES USED, PRODUCED, OR OTHERWISE PRESENT

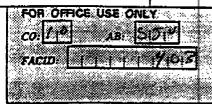
FORM S-UP

			r
FACIL	ίΤι.	NAME	t

FRESNO INDUSTRIAL ASPHALT	USTRIAL ASPHALT	7
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PLEASE COPY THIS FORM AS MANY TIMES AS NECESSARY FOR YOUR FACILITY. PLEASE READ THE INSTRUCTIONS BEFORE COMPLETING THIS FORM.

USE THIS FORM TO REPORT SUBSTANCES IN APPENDIX A-II WHICH ARE USED, PRODUCED, OR OTHERWISE PRESENT.



PLEASE INDICATE (YM) UNDER THE APPROPRIATE CATEGORIES IUSE, PRODUCTION, OR OTHER PRESENCE WITHIN YOUR FACILITY) OF ANY SUBSTANCEIS LISTED IN APPENDIX A-11. "USED" REFERS TO SUBSTANCES WHICH ARE INGREDIENTS IN ANY ACTIVITY OR PROCESS AT YOUR FACILITY. "PRODUCED" REFERS TO SUBSTANCES WHICH ARE THE RESULT OF ANY ACTIVITY OR PROCESS TAKING PLACE IN YOUR FACILITY. "OTHERWISE PRESENT" REFERS TO SUBSTANCES PRESENT IN ANY OTHER WAY IN AN ACTIVITY OR PROCESS, SUCH AS BY-PRODUCTS OR REACTION INTERMEDIATES WHICH APPEAR TEMPORARILY DURING PROCESSING. PLEASE SPECIFY THE NATURE OF THE PRESENCE OF THE SUBSTANCE.

ALSO USE THIS FORM TO REPORT SUBSTANCES IN APPENDIX A-I WHICH ARE PRESENT BELOW THE APPLICABLE DEGREE OF ACCURACY.

LISTED SUBSTANCE	USED	PRODUCED -		SPECIFY)	_	
12/0	(Y)	(N)	(N) ASPHALTIC	STORAGE AND CONCRETE LOAD	<u>ہر</u>	- 石
	()	()	(·) SiLos	AND TRUCKS.	_	Ť
7/576	(/)	(N)	(~)		_	
56553	(y)	(N)	(N)	· /		
205-992	(y)	(4)	(~)			
207389	(y)	(~)	(~)	.,,	_	
50328	(y)	(\(\sigma \)	(h)			
53703	(y)	(v)	(⁽ /)	1		
/93395	(y)	(4)	(h)	"	_	
912 03	(y)	(~)	(¹ / ₂)	"		
· · · · · · · · · · · · · · · · · · ·	()	()	()	·	_	
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	()	()	()	·	_	
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				•		

NAME: Attorine ASS ioun

DATE: 5-25-96 ARE/S-UP/89089

* T	
emission vear supplemental process parameter form supplemental process parameter form stationary combustion	S-CME
COMPANY NAME FRESHO INDUSTRIAL ASPHALT CONTINUE	ata in Change Landing (Section 1)
DEVICE ID: 3	101101 6 (0)
PLEASE COPY THIS FORM AS MANY TIMES AS NECESSARY FOR YOUR FACILITY. PLEASE READ THE INSTRUCTIONS BEFORE COMPLETING THIS FORM.	
FUEL & FEEDSTOCK COMPOSITION	
UNCONVENTIONAL FUELS & FEEDSTOCKS:	DEVICE ID
INDICATE WITH A CHECKMARK ANY UNCONVENTIONAL FUELS OR FEEDSTOCKS USED IN YOUR FACILITY AT THE INDICATED ABOVE. DESCRIBE BRIEFLY THE NATURE OF SUCH FUEL OR FEEDSTOCK IN THE SPACE PROVIDED. ALSO SUMMARIZE THIS FEEDSTOCK INFORMATION IN THE DATA FIELD, "FUEL TYPE/OTHER PROCESS INFO" ON C	
MUNICIPAL WASTE	
HOSPITAL WASTE	
HAZARDOUS WASTE	<u> </u>
WASTE OIL	
WASTE SOLVENT	
AGRICULTURAL DEBRIS	
TIRES	
OTHER PLEASE SPECIFY: DIESEL	
REPORT USE OF AUXILIARY FUEL WITH THESE FEEDSTOCKS IN ACCORDANCE WITH THE STATIONARY COMBUST INSTRUCTIONS FOR THIS FORM, S-CMB.	ION REPORTING
FUEL & FEEDSTOCK ANALYSIS	
COMPLETE THIS PART FOR EACH FUEL AND FEEDSTOCK USED AT THE DEVICE ID INDICATED ABOVE FOR WHICH AMALYSIS WAS PERFORMED.	1 A FUEL
(IN WEIGHT %): SULFUR:	
IIN PARTS-PER-MILLION BY WEIGHT (PPMW) OR MG/KG:	
. CHROMIUM VI. 1 1 1 1 1 1 PHOSPHORUS: 1	
ARSENIC: WI-101111 COPPER: 1-191111 RADIONUCLIDES:	1.12.
BERYLLIUM: N.D.	
BROMINE: 1 1 1 1 1 1 1 LEAD: 3: 41 1 1 1 1 ZINC: 319: 51 :	<u>i 1</u>
CADMIUM: 01-141 1 MANGANESE 01-141 1 1 1	

NICKEL: M. DIOI I I ARB/S-CMB/89100 Antoine Assider NAME: __ DATE: ___

REPORT EMISSIONS IN SECTION 2 OF CORE FORM PRO

MERCURY: 01101017111

CHLORINE: NID 11 1 : :

DOCUMENTATION SUPPORTING THE EMISSIONS CALCULATIONS

The attached documentation consists of printouts from the Lotus 123 spreadsheet program used to calculate emissions from this facility. The printouts are comprehensive in that they show the exact equations used, the input parameter values, as well as the calculated output values for each reported process.

Marin Makildic LIGHT / ECHECT

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Device #1: Aggregate Transfer Points No. of Devices: 3 Control Equipment: Wet Supression Estimation Method: Emission Factor Gnrl Subst Quantfd: Particulates Yearly Emis Est Equatn: Ay*Ef*(1-Ce)*Sf/1,000,000 Hourly Emis Est Equatn: Am*Ef*(1-Ce)*Sf/1,000,000 Max Hr Agg Rate Equatn: Am = Ay*(Pm/Py) Parameter Symbols/Names Values Ay = Total Aggregate Material Annual Process Rate 1.19E+06 tons/yr 8 hrs/day Hd = Hours per Day Dw = Days per Week 5 days/wk Wy = Weeks per Yéar 52 wks/yr Ah = Aggregate Average Hourly Process Rate 571.8908653 tons/nr Am = Aggregate Maximum Hourly Process Rate 1575.089017 tons/hr Fy = Total Yrly Asphaltic Concrete Production Rate 415369 tons/yr Pm = Max Hrly Asphaltic Concrete Production Rate 550 tons/hr £f = Gnrl Subs PM10 Emis Factor (uncontrolled)* 0.015 lb/ton tle = Control Efficiency 70% 3f = Speciation Factor (see balow) ppm * Note: Based on EPA AF-42 p.8.19.1-3 & EPA Particle Size Distribution Interim Report (July '86) p.C.2-10. Speciation Emission Annual Avg Hourly Max Emittent Factor Factor Emissions Emissions Species Name (ppm) (lbs/ton) (lbs/yr) (lbs/hr) _____ Crystalline Silica 156175 0.0002342625 278.6629744 0.368984290 Total Particulates 1,000,000 0.0015 1784.2995 2.36263\$525 ١٠

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Device #2: Dryer/Mixer System
        No. of Devices: 1
     Control Equipment: Baghouse
     Estimation Method: Source Test
Yearly Emis Est Equatn: Py*Ef
Hourly Emis Est Equatn: Pm*Ef
       Emission Factor: Ef = Et/Pt
Parameter Symbols/Names
                                                     Values
Py = Total Yearly Asphaltic Concrete Production Rat
                                                   4.15E+05 tons/yr
Pm = Max Hrly Asphaltic Concrete Production Rate
                                                       550 tons/hr
Ef = Et/Pt = Emission Factor
                                                (see below) lbs/ton
Pt = Hourly Production Rate during Source Test
                                                       475 tons/hc
Et = Tested Emission Rate
                                                (see below) lbs/hr
6t = Tested Actual Gas Flow Rate of Stack
                                                     55706 cfm
Gn = Gt*P/(Pt*H) = Normal Actual Gas Flow Rate
                                                23419.58048 cfm
T = Tested Stack Gas Temperature
                                                       316 degree F
                                       ------
                       Tested Emis
                                                 Annual Avg Hourly Max
                                       Emission \
                       Rate ("Et")
       Emittent
                                         Factor
                                                  Emissions
                                                            Emissions
     Species Name
                          (lbs/hr)
                                       (lbs/ton)
                                                   (lbs/yr)
                                                              (1bs/hr)
Heavy Metals:
  Arsenic
                         1.200E-04
                                    0.0000002526 0.104935326 0.00013B947
   Beryllium
                           N.D.
                                                                     C:
   Cadmium
                         1.200E-04
                                    0.0000002526 0.104935326 0.00013B947
   Chromium (total)
                          N.D.
   Chromium (hexavalent)
                           N.D.
                                              O
                                                         ()
                                                                     0
   Copper
                           N.D.
                                              0
                                                         Ō
                                                                     Ŏ
   Lead
                         2.900E-04
                                    0.0000006105 0.253593705 0.000335789
   Manganese
                                    0.0025263158 1049.353263 1.389473684
                         1.200E+00
                                    0.000000074 0.003060613 0.000004052
   Mercury
                         3.500E-06
   Nickel
                           N.D.
                                              Ò
                                                         Ü
                                                                     O
   Selenium
                           N.D.
                                              0
                                                                     0
   Zinc
                          1.500E-02
                                    0.0000315789 13.11691578 0.017348421
PAHS:
   Total
                         B.400E-03
                                    0.0000176842 7.345472842 0.009726315
   Benz[a]anthracene ·
                           N.D.
                                              Ö
                                                        · O
                                                                     0
   Benzo[b]fluranthene
                          N.D.
                                                         O
                                                                     O
   Benzo[k]fluoranthene
                           N.D.
                                              O
                                                          0
                                                                     O
   Benzo(a)pyrene
                           N.D.
                                              0
                                                         O
                                                                     O
   Dibenzo[a.h]anthracene
                           N.D.
                                              0
                                                          O.
                                                                     Û
   Indeno[1,2,3,-cd]pyren
                           N.D.
                                                                     O
   Naphthalene
                         5,900E-03 0.0000124211 5.159320210 0.006831578
Hydrogen Sulfide
                           N.D.
Formaldehyde
                          3.200E-01
                                    0.0006736842 279.8275368 0.370526315
Benzene
                           N.D.
                                              0
                                                         Ō.
                                                                     O.
Toluene
                           N.D.
                                              Ō
```

Xylene	N.D.	Q	o	
Methyl Chloroform	N.D.	0	. 0	1
<u>~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~</u>	๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛๛		·~~~~~~~~~~	40000000
Estimation Method: Yearly Emis Est Equatn: Hourly Emis Est Equatn:	Py≠Ef≠Sf	or + Analysi	s of Fugitive	Dust Samp
Parameter Symbols/Names		:=	Values	
Ef= Particulate Emission Sf = Speciation Factor		source test)	0.00573 (see below)	
Emittent Species Name	Speciation Factor	Emission Factor	Annual Avg Emissions	Hourly Ma Emission
	(mqq)		(lbs/yr)	
Crystalline Silica	250,000	0.0014325	595.0160925	0.7 87 87

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Device #3: Hot Oil Heater

No. of Devices: 1 Control Equipment: None

Cationtine Mathod: Source Toct

Estimation Method: Source Test

Yearly Emis Est Equatn: Fy*Ef

Hourly Emis Est Equatn: (Yearly Emissions)/H

Emission Factor: Ef = Et/Ft

Parameter Symbols/Names	Values		
Fy = Total Yearly Fuel Consumption H = Total Yearly Hours of Operation Fh = Average Hourly Fuel Consumption Ef = Et/Ft = Emission Factor Ft = Hourly Fuel Consumption during Source Test Et = Tested Emission Rate St = Tested Actual Gas Flow Rate of Stack Gn = Gt*Fy/(Ft*H) = Normal Actual Gas Flow Rate	5,556 4.499640028 (see below) 2.5 (see below)	lbs/gal gal/hr lbs/hr cfm	
T = Tested Stack Gas Temperature		degree	F

Emittent Species Name	Tested Emis Rate ("Et") (1bs/hr)	Emission Factor (lbs/gal)	Annual Avg Emissions (lbs/yr)	Hourly Max Emissions (1bs/hr)
PAHs:				.
Total	6,300E-05	0.0000252	0.63	0.000113390
Benzlalanthracene	N.D.	Q	0	0
Benzotblfluranthene	2.500E-07	0.0000001	0.0025	o.000p0045
Benzo[k]fluoranthene	N.D.	Q	0	, o
Benzo[a]pyrene	N.D.	. 0	Q	0
Dibenzola,hlanthracen	e N.D.	Q.	Ó	0
Indeno[1,2,3,-cd]pyre		Q	0	Ü
Naphthalene	4.200E-05	0.0000148	0.42	0.000075594
Dioxins	Ņ.D.	0	o	. 0
Furans	N.D.	, 0	0	0
Formaldehyde	6.800E-02	0.0272	680	0.122370208
Benzene	N.D.	٥	0	0

Estimation Method: Fuel Analysis

Yearly Emis Est Equatn: Fy*k*C

Hourly Emis Est Equatn: (Yearly Emissions)/H

Emission Factor: k*C

Parameter Symbols/Names	Values
k = 3.785*2.205/1,000,000	0.000008345 (1/gal)(1b/m
Fy = Yearly Fuel Usage	25000 gal/y†

C = Concentration	in	fuel
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(see helow) mg/liter

		.===========	*=========	===
Emittent Species Name	Concentratn in Fuel (mg/liter)	Emission Factor (lbs/gal)	Annual Avg Emissions (lbs/yr)	- /
Chlorine	N.D.	0	Q.	0
Heavy Metals:				
Arsenic	N.D.	O	Q	0
Beryllium	N.D.	Ö	O.	0
Cadmium	3.500E-01	0.0000029211	0.073026843	0.000013143
Chromium (total)	6.000E-01	0.0000050076		
Copper	1.600E+00	0.0000133535	0.333837	0.000060085
Lead	2.900E+00	0.0000242032	0.605079562	0.000108905
Manganese	- 3.000E-01	0.0000025038	0.062594437	0.000011266
Mercury	6.000E-03	0.0000000501		
Nickel	N.D.	Q.	O	O.
Seienium	N.D.	0	Ö	i è
Zinc	3.340E+01	0.0002787539	6.968847375	0.001254292
	(pCuries/l)	(mCuries/gal)	(Curies/yr)	(mCurries/h
Radionuclides	N.D.	, o	0	0

FUEL ANALYSIS AND STACK TEST REPORTS

REPORT OF AB2588 AIR POLLUTION SOURCE TESTING AT INDUSTRIAL ASPHALT FRESNO, CALIFORNIA

Conducted at:

INDUSTRIAL ASPHALT FRESNO, CALIFORNIA

Conducted on:

May 22-24, 1990

Submitted on:

August 23, 1990

Prepared by:

ENGINEERING-SCIENCE, INC. 75 N. Fair Oaks Avenue P.O. Box 7101 Pasadena, California 91109

REPORT OF AB2588 AIR POLLUTION SOURCE TESTING AT INDUSTRIAL ASPHALT FRESNO, CALIFORNIA

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REPORT OF AB2588 AIR POLLUTION SOURCE TESTING AT INDUSTRIAL ASPHALT FRESNO, CALIFORNIA

SECTION I INTRODUCTION

On May 22-24, 1990, Engineering-Science, Pasadena, (ES), conducted air pollution source tests at the Industrial Asphalt facility in Fresno, California. The sources tested were the exhausts of the dryer mixer and hot oil heater plant mixing operations. The testing was conducted to meet the conditions stipulated in California Assembly Bill 2588 (AB 2588). The plant is permitted by the Fresno County Air Pollution Control District (FCAPCD). The test effort was coordinated by Mr. Dan Olivera of Industrial Asphalt. The ES testing team was comprised of Messrs. Mike Edwards (team leader), Rani Sekhon, Tony King, and Greg Burke.

The source testing program included the determination of emission rates of trace elements, poly-aromatic hydrocarbons (PAH), organics, formaldehyde, and hydrogen sulfide for the dryer/mixer stack. For the hot oil heater stack, emission rates for poly-aromatic hydrocarbons, dioxins, dibenzofurans, benzene, and formaldehyde were determined. Fuel analysis included trace elements, chloride, and radionuclides. The tests were conducted in accordance with California Air Resources Board (CARB), and Environmental Protection Agency (EPA) published test procedures.

SECTION II EQUIPMENT AND PROCESS DESCRIPTION

Industrial Asphalt of Fresno, California, operates a drum mixing asphalt batch plant for production of various grades of road asphalt. The facility is located in a large drainage wash, adjacent to their quarry. All gravels and sands necessary for aphalt production are generated on-site.

The process starts with differentiation of quarry aggregate into different grades of gravel and sand. These materials are stored in large piles at the facility, and as needed, are loaded into hoppers by skip loader for use by the plant.

Beneath the hoppers are belt conveyors, all of which feed a common conveyor belt. By adjusting the belt speeds for the individual hoppers, mixtures of different gravel sizes, sand and rock dust of desired proportions are created. The entire process is controlled by a central computer.

The common conveyor belt feeds materials through a knockout screen (protection against large stones) and into the top of the rotary drum. The materials are dried and heated at the top of the mixing drum by a propane flame. As the mixture falls through the last 15 feet of the drum, it is coated by 300°F asphalt oil. The asphalt which emerges from the mixing drum is carried by drag-slat up into storage silos, from where it is eventually poured into trucks.

To prevent the asphalt oil and asphalt mix from solidifying, a burner fired by low sulfur diesel heats circulating oil. The circulating oil maintains the temperature of the three 125-ton asphalt oil storage tanks and five asphalt mix silos at 300°F.

At maximum capacity, this facility can produce 650 tons of asphalt mix per hour. The plant load fluctuates constantly depending on the contractor demand, and can vary anywhere from complete shutdown to maximum capacity.

SECTION III

TESTING METHODOLOGY

Exhaust Gas Velocity and Moisture Determination

The exhaust gas flow rate was determined using an S-Type pitot tube. A Type-K thermocouple (chromel-alumel) connected to an Omega Model 601, digital temperature readout was used to determine the exhaust gas temperature. Carbon dioxide (CO₂) and oxygen (O₂) used in determining the molecular weight of the exhaust gases were determined by Fyrite analysis of integrated bag samples collected simultaneously with the wet impingement sampling trains. Moisture content was determined gravimetrically by the weight gain of the impingers from the particulate train in accordance with EPA Method 4.

Testing for cyclonic flow was conducted in accordance with EPA Reference Method 1. Absence of cyclonic flow was verified by rotating an S-Type pitot tube so that the planes of the face openings of the pitot tube are perpendicular to the stack cross-sectional plane. This is known as the "0 degree reference" or "null position". A zero reading should be obtained from the manometer. If the manometer did not read zero, the pitot tube was rotated up to a 90° yaw angle or until a zero reading was obtained. The angle of rotation from the initial position was recorded to the nearest degree. The source was considered free of cyclonic flow.

Trace Elements

Trace elements that were analyzed included As, Be, Cd, Cr, Cu, Hg, Pb, Mn, Ni, Se, Zn. Each sample was collected isokinetically using an EPA proposed multiple metals sampling train. The train was configured similar to the sampling train used for determining particulates (Method 5) except the first and second impingers contained nitric acid (HNO₃) in a hydrogen peroxide (H₂O₂) solution. The third impinger was an empty modified Greenberg-Smith. The fourth impinger contained an acidified potassium permanganate (KMnO₄) solution. The fifth impinger contained approximately 400 grams of indicating silica gel. A glass fiber filter was located between the fourth and fifth impingers. Sampling train components were

recovered in separate HNO₃ and KMnO₄ fractions. After the testing, the samples were analyzed by Inductively Coupled Argon Plasma Mass Spectrometry (ICAP-MS).

Total and Hexavalent Chromium

Total and Hexavalent Chromium sampling was conducted in accordance with CARB Method 425. The sampling technique is the same as that for CARB Method 5 for particulates except the impinger solution is sodium bicarbonate. A sample aliquot was analyzed for Cr⁶⁺ by a colorimetric technique and an aliquot was analyzed for total Cr by AAS with graphite furnace.

Audit metal, Be

The audit metal, Beryllium, was sampled as per EPA Method 104. The method is similar to EPA5 except for analysis.

Poly-aromatic Hydrocarbons, Dioxins, and Dibenzofurans

A modified EPA Reference Method 5 sampling train was used for the determination of Dioxins (PCDD), Dibenzofurans (PCDF), and poly-aromatic hydrocarbons (PAH) which are in accordance with CARB Method 428 and 429. The train design was based on an EPA validated emission collection system with the addition of an adsorbent cartridge of XAD-2 resin to collect vaporous emissions for semi-volatile organics. This train was operated for a period of 4 hours or more during each run and samples were collected using isokinetic sampling techniques. The train recovery was modified from Method 5 procedures to include a sample wash with deionized water followed by a sample wash each with acetone, methylene chloride, and toluene.

A quartz glass sampling probe liner was used. Organics were collected by the adsorbent trap containing a precleaned cartridge of XAD-2 Resin. The resin cartridges were precleaned by California Analytical Laboratory in Sacramento, California at least two weeks prior to the field testing. This glass trap was located in the sample line downstream of a heated filter holder and upstream of the first impinger. The module housing the trap was jacketed, with cold water circulating to maintain an outlet temperature below 60°F. Aluminum foil was wrapped around the sorbent tube to minimize any possible sample reactions caused by ultraviolet

light. A glass Hempal-type condenser was located between the filter and the XAD-2 cartridge to ensure that cool stack gas was entering the adsorbent trap.

All solvents used for preparing the sampling train for testing and field sample recovery were stored in glass bottles and were spectrographic grade. The train components that were to be in contact with the sample were handled with clean, bare hands. These components were free of all potential interfering materials, especially silicone grease.

The probe, sample line wash, and glass condenser were rinsed with deionized water, acetone, methylene chloride, and toluene during each rinse. All probe, filter, connecting tubing, and impinger washings were collected in precleaned glass containers. The sample train was separated into front and back halves.

Three composite samples from each sampling run were submitted to the laboratory for analysis. A composite sample of the "front half" was comprised of the nozzle/probe wash, filter wash, and filter. The "back half" composite sample was comprised of the filter back half wash, condenser wash, flexible line wash and XAD-2 cartridge. Another composite consisted of the impinger contents and associated washes. The samples were sealed, labelled and shipped with Chain of Custody forms to California Analytical Laboratories in Sacramento, California for analysis. The dioxins and dibenzofurans were analyzed, as specified by CARB Methods 428 and 429, in accordance with EPA Method 8280. The PAH's were analyzed by gas chromatography mass spectrometery (GC/MS) in accordance with EPA Method 8270. The analysis for PCDD and PCDF included the following compounds:

<u>PCDD</u>	PCDF
Mono-CDD	Mono-CDF
Di-CDD	Di-CDF
Tri-CDD	Tri-CDF
Tetra-CDD	Tetra-CDF
Penta-CDD	Penta-CDF
Hexa-CDD	Hexa-CDF
Hepta-CDD	Hepta-CDF
Octa-CDD	Octa-CDF

Concurrent with the laboratory analyses, the following poly-aromatic hydrocarbons (PAH's) were determined:

PAH's

Fluoranthane Pyrene Benzo (a) anthracene Chrysene Benzo (a) pyrene Acenaphthene Naphthalene Benzo (b) fluoranthene Benzo (k) fluoranthene Acenaphthylene Anthracene Benzo (g, h, i) perylene Fluorene Phenathrene Dibenzo (a, h) anthracene Indeno (1,2,3-cd) pyrene

The samples that were submitted to the laboratory included a sample train blank which was collected on site. A sampling train was prepared as if it were to be used on the stack, but without being used, the train was washed with the appropriate solvents which were collected in the respective containers.

Benzenes

Benzene, dichlorobenzenes, ethylene dichloride and ethylene dibromide samples were collected in accordance with CARB Method 410A.

Three integrated samples were collected into evacuated summa polished canisters. The analysis were performed using a gas chromatrograph (GC) equipped with a photoionization detector (PID).

Hydrogen Chloride

Using CARB method 421, hydrogen chloride samples were collected with wet impingement sampling trains. The first and second impingers were charged with a solution of sodium bicarbonate and sodium carbonate. The samples were collected through a stainless steel probe connected to a teflon sampling line. Analysis of the samples were conducted by ion chromatrography with conductivity detection.

Hydrogen Sulfide

Samples were collected and analyzed in accordance with EPA method 11. A midget impinger train charged with a scrubbing solution of hydrogen peroxide and an absorbing solution of cadmium sulfate.

Formaldehyde

Samples were collected in accordance with CARB Method 430. Three runs each 2 hours in duration were conducted on the stack outlet.

The sampling train consisted of a Teflon lined probe connected to three midget impingers in series. The first two impingers contained 10 mls of 2,4-dinitrophenyl-hydrazine (DNPH) and the third impinger was empty. A preweighed silica gel cartridge was attached between the third impinger and the pump to prevent moisture entering the pump and for use in determining the moisture content of the stack exhaust gas. Samples were analyzed by high performance liquid chromatography.

Radionuclides Sampling

Gross alpha and beta radioactivity were measured using Methods 601 and 602 of the Intersociety Committee "Methods of Air Sampling and Analysis", third edition. Samples were collected once per hour and composited for each source test run. A total of three composite samples were collected.

SECTION V RESULTS

The test results are presented in Tables 1-9. Manganese emissions are not deemed reliable because of the possible migration of the highly concentrated permanganate solution in the EPA multiple metal train.

Table 2
Summary of Poly-Aromatic Hydrocarbon Emissions Data
Dryer/Mixer Stack

Parameter	Run #1 lb/hr.	Run #2 lb/hr.	Run #3 lb/hr.	Average lb/hr
Naphthalene	5.6 x 10 ⁻³	5.0 x 10 ⁻³	7.0 x 10 ⁻³	5.9 x 10 ⁻³
Acenaphthylene	3.5 x 10 ⁻⁵	4.9 x 10 ⁻⁵	6.4 x 10 ⁻⁵	4.9 x 10-5
Acenaphthene	2.4 x 104	2.1 x 104 (1777)	3.6 x 104	2.7 x 104
Fluorene	3.0 x 10-4	3.5 x 104	5.1 x 10 ⁻⁴	3.9 x 10-4
Phenanthrene	6.0 × 10-4	3.5 x 10 ⁻³	9.8 x 10 ⁻⁴	1.7 x 10 ⁻³ % 60 × 1
Anthracene	2.1 x 10 ⁻⁵	3.2 x 10 ⁻⁵	5.1 x 10 ⁻⁵	3.5 x 10.5 - 10.5
Fluoranthene	5.0 x 10.6	< 7.3 x 10-6	1.6 x 10 ⁻⁵	< 9.4 x 10-6
Pyrene	8.2 x 10-6	< 1.3 x 10 ⁻⁵	2.6 x 10 ⁻⁵	< 1.6 x 10 ⁻⁵
Benzo (a) anthracene	< 2.0 x 10 ⁻⁷	< 4.9 x 10-6	< 3.3 x 10 ⁻⁶	< 2.8 x 10.6
Chrysene	2.3 x 10-6	< 6.5 x 10-6	< 4.1 x 10 ⁻⁶	< 4.3 x 10 ⁻⁶ 2 ** ** **
Benzo (b) fluoranthene	4.5 x 10 ⁻⁵	3.2 x 10 ⁻⁵	< 5.4 x 10 ⁻⁶	< 2.8 x 10-5
Benzo (k) fluoranthene	< 1.5 x 10-6	< 8.4 x 10-6	3.3 x 10 ⁻⁵	< 1.4 x 10-5
Benzo (a) pyrene	< 2.5 x 10 ⁻⁷	< 1.0 x 10 ⁻⁵	< 4.4 x 10 ⁻⁶	< 4.9 x 10-6
Dibenz (a,h) anthracene	< 3.2 x 10-8	< 4.6 x 10-6	< 2.8 x 10 ⁻⁶	< 2.5 x 10 ⁻⁶
Benzo (g,h,i) perylene	< 8.7 x 10.8	< 1.5 x 10 ⁻⁵	< 7.2 x 10 ⁻⁶	< 7.4 x 10.6 L/C = 5
Indeno (1,2,3-cd) pyrene	< 4.7 x 10.8	< 6.5 x 10 ⁻⁶	< 2.8 x 10-6	< 3.1 x 10 ⁴

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Summary of Organics, Formaldehyde, and Hydrogen Sulfide Emission Data Dryer/Mixer Stack Table 3

Parameter	Run #1 1b/hr.	Run #2 lb/hr.	Run #3 lb/hr.	Average Lib/hr
Benzene	< 2.6 x 10 ⁻²	< 2.6 x 10 ⁻²	< 2.6 x 10-3	<2.6 x 10-2 .00 ···
Toluene	1.0 x 10-1 > 1 \ \ 1.0 x 10-1 \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	1.3 x 10 ⁻¹	< 2.6 x 10 ⁻²	< 8.5 x 10 ⁻² (2) 18 × 5
Xylene	< 2.6 x 10 ⁻²	< 2.6 x 10 ⁻²	< 2.6 x 10 ⁻²	< 2.6 x 10 ⁻²
Methyl Chloroform	< 2.6 x 10 ⁻²	< 2.6 x 10 ⁻²⁷ (17)	4.2 x 10-2 1.7 1 10 5	< 3.1 x 10 ⁻² (p. 1 per
Formaldehyde	3.7 x 10 ⁻²	5.0 x 10-1	4.2 x 10-1	3.2 x 10.1 × E = 2.5%
Hydrogen Sulfide	< 2.1 x 10 ⁻²	< 3.0 x 10-2	< 3.4 x 10 ⁻²	< 2.8 x 10 ⁻² 3.4 x P ⁻²

Table 4
Summary of Poly-Aromatic Hydrocarbon Emission Data
Hot Oil Heater Stack

Parameter	Run #1 1b/hr.	Run #2 lb/hr.	Run #3 lb/hr.	Average lb/hr
Naphthalene	5.8 x 10 ⁻⁵	4.2 x 10 ⁻⁵	2.7 x 10 ⁻⁵	4.2 x 10 ⁻⁵
Acenaphthylene	6.3 x 10 ⁻⁷	5.1 x 10-7	3.6 x 10-7	5.0 x 10-7
Acenaphthene	1.4 x 10.6	1.3 x 10-6	1.3 x 104	1.3 x 10 ⁻⁶
Fluorene	9.4 x 104	3.3 x 104	4.8 x 10-6	5.8 x 104
Phenanthrene	1.7 x 10 ⁻⁵	1.0 x 10 ⁻⁵	1.0 x 10 ⁻³	1.2 x 10 ⁻⁵
Anthracene	4.0 x 10 ⁻⁷	3.4 x 10-7 11 11 P	6.0 x 10-7 20 1 1 2 1	4.5 x 10-7
Fluoranthene	1.3 x 10-7	7.1 x 10 ⁻⁸	1.3 x 10-7	12 10 rate
Pyrene	9.7 x 10-8	7.1 x 10 ⁻⁸	6.8 x 10 ⁻⁸	7.9 x 10-8
Benzo (a) anthracene	< 9.5 x 10.9		< 1.9 x 10 ⁻⁸	$< 1.2 \times 10^{-8}$
Chrysene	< 8.5 x 10.9	< 7.4 x 10.9 % (1) 1 × 7.4 x 10.9 % (1) 1 × 10.0 €	< 1.9 x 10 ⁴ = 100 x 10 x	$< 1.2 \times 10^{-8}$
Benzo (b) fluoranthene	3.7 x 10-7	1.5 x 10-7	2.4 x 10-7	2.5 x 10 ⁻⁷
Benzo (k) fluoranthene	$< 1.2 \times 10^{-8}$	$< 1.0 \times 10^{-8}$	< 9.7 x 10.9	< 1.1 x 10.8
Benzo (a) pyrene	< 6.3 x 10 ⁻⁹	< 5.7 x 10.9	< 8.4 x 10 ⁻⁹	< 6.8 x 10.9
Dibenz (a,h) anthracene	$< 1.2 \times 10^{-9}$	< 2.4 x 10.9	< 1.4 x 10.9	< 1.7 x 10.9
Benzo (g,h,i) perylene	$< 2.9 \times 10^{-9}$	< 2.8 x 10.9	< 5.2 x 10.9	< 3.6 x 10° V
Indeno (1,2,3-cd) pyrene	< 1.5 x 10.9	< 3.1 x 10.9	< 4.2 x 10.9	< 2.9 x 10 ⁻⁹

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Table 5
Summary of Polychlorinated Dibenzofurans Emissions Data
Hot Oil Heater Stack

Parameter	Run #1 lb/hr.	Run #2 lb/hr.	Run #3 lb/hr.	Average lb/hr
TCDFs (total)	< 2.8 x 10 ⁻¹²	6.4 x 10 ⁻¹²	1.7 x 10·11	< 8.7 x 10 ⁻¹²
2,3,7,8, -TCDF	< 1.1 x 10·12	$< 2.0 \times 10^{12}$	< 4.2 x 10 ⁻¹²	< 2.4 x 10 ⁻¹²
PCDFs (total)	< 9.8 × 10-13	$< 1.0 \times 10^{12}$	2.6 x 10-12	< 1.5 x 10 ⁻¹²
1,2,3,4,7,8 -PCDF	< 9.8 x 10-13	$< 1.0 \times 10^{-12}$	$< 2.0 \times 10^{-12}$	$< 1.3 \times 10^{-12}$
2,3,4,7,8 -PCDF	< 1.6 x 10-12	$< 1.7 \times 10^{12}$	$< 2.0 \times 10^{-12}$	$< 1.8 \times 10^{-12}$
HxCDFs (total)	< 6.6 x 10-13	2.7 x 10 ⁻¹²	1.2 x 10 ⁻¹¹	< 5.1 x 10 ⁻¹²
1,2,3,4,7,8 -HxCDF	$< 1.9 \times 10^{-12}$	$< 2.0 \times 10^{-12}$	< 5.5 x 10 ⁻¹²	< 3.1 x 10 ⁻¹²
1,2,3,4,6,7,8 -HxCDF	$< 7.6 \times 10^{-13}$	< 8.1 x 10 ¹³	$< 2.0 \times 10^{-12}$	$< 1.2 \times 10^{-12}$
2,3,4,6,7,8 -HxCDF	$< 7.6 \times 10^{13}$	< 8.1 x 10 ⁺¹	< 2.0 x 10 ⁻¹²	$< 1.2 \times 10^{-12}$
1,2,3,7,8,9 -HxCDF	$< 7.6 \times 10^{-13}$	$< 8.1 \times 10^{-13}$	$< 2.0 \times 10^{-12}$	$< 1.2 \times 10^{-12}$
HpCDFs (total)	$< 1.6 \times 10^{-12}$	1.7 x 10 ⁻¹¹	5.5 x 10 ⁻¹¹	< 2.5 x 10 ⁻¹¹
1,2,3,4,6,7,8 -HpCDF	$< 3.8 \times 10^{-12}$	$< 6.7 \times 10^{-12}$	2.1 x 10 ⁻¹¹	< 1.1 x 10 ⁻¹¹
1,2,3,4,7,8,9 -HpCDF	$< 1.9 \times 10^{-12}$	< 6.7 x 10 ⁻¹²	< 3.2 x 10 ⁻¹²	$< 3.9 \times 10^{-12}$
OCDF	< 5.0 x 10-12	9.4 x 10·12	7.8 x 10-11	< 3.1 x 10 ⁻¹¹

Tuble 6
Summary of Dioxin Emissions Data
Hot Oil Heater Stack
Industrial Asphalt
Fresno, California
May 22-24, 1990

Parameter	Run # 1 lb/hr.	Run #2 16/hr.	Run #3 lb/hr.	Average lb/hr
TCDDs (total)	< 1.9 x 10 ⁻¹²	< 2.1 x 10-12	< 2.0 x 10-12	< 2.0 x 10-12
2,3,7,8 -TCDD	< 1.5 x 10 ⁻¹²	< 9.8 x 10 ¹³	$< 1.7 \times 10^{-12}$	$< 1.4 \times 10^{-12}$
PeCDDs (total)	$< 2.0 \times 10^{-12}$	$< 2.1 \times 10^{-12}$	< 3.1 x 10 ⁻¹²	< 2.4 x 10 ⁻¹²
1,2,3,7,8-PeCDD	$< 2.0 \times 10^{-12}$	< 2.1 x 10 ¹²	< 3.1 x 10 ⁻¹²	< 2.4 x 10 ⁻¹²
HxCDDs (total)	4.7 x 10-12	1.4 x 10-11	2.8 x 10 ⁻¹¹	1.6 x 10 ¹¹
1,2,3,6,7,8 -HxCDD	< 1.6 x 10 ⁻¹²	< 1.8 x 10 ¹²	< 4.5 x 10 ⁻¹²	< 2.6 x 10 ⁻¹²
1,2,3,7,8,9 -HxCDD	< 1.6 x 10 ⁻¹²	2.6 x 10-12	< 4.5 x 10 ⁻¹²	< 2.9 x 10 ⁻¹²
1,2,3,4,7,8 -HxCDD	$< 1.6 \times 10^{-12}$	2.1 x 10 ⁻¹²	$< 4.5 \times 10^{-12}$	$< 2.7 \times 10^{-12}$
HpCDDs (total)	$< 7.2 \times 10^{-12}$	< 7.8 x 10-12	1.4 x 10-10	< 5.2 x 10-11
1,2,3,4,6,7,8 -HpCDD	$< 7.2 \times 10^{-12}$	1.2 x 10 ⁻¹¹	9.6 x 10 ⁻¹¹	< 3.8 x 10-11
OCDD	< 5.0 x 10 ⁻¹¹	9.4 x 10 ⁻¹¹	1.1 x 10-9	< 4.2 x 10-10

Table 7 Summary of Benezene and Formaldehyde Emissions Data Hot Oil Heater Stack Industrial Asphalt Fresno, California May 22-24, 1990

Parameter	Run #1 lb/hr.	Run #2 lb/hr.	Run #3 lb/hr.	Average lb/hr
Benzene	< 3.8 x 10-4	< 3.8 x 10-4	< 3.8 x 10-4	< 3.8 x 10 ⁻⁴
Formaldehyde	4.0 x 10 ⁻²	5.5 x 10 ⁻²	1.1 x 10 ⁻¹	6.8 x 10-2
	1.6 -2	<u>. </u>	44-2	2.7 X/V-Z

Table 8 Summary of Chloride and Trace Element Data Fuel Oil to Hot Oil Heater Stack Industrial Asphalt Fresno, California May 22-24, 1990

Parameter	mg/l
As	< 0.1
Ве	< 0.1
Cd	0.35
CI	< 45.0
Cr	0.6
Cu	1.6
Hg	0.006
Mn	0.3
Ni	< 0.2
Рь	2.9
Se	< 0.2
Zn	33.4

Table 9
Summary of Radionuclides Data
Fuel Oil to Hot Oil Heater Stack
Industrial Asphalt
Fresno, California
May 22-24, 1990

Pa	arameter	pico curie/l	
Gro	oss Alpha	0 ± 7	
Gro	oss Beta	0 ± 13	
Ga	mma Scan:		
K ⁴⁰		$(8.4 \pm 7.8) \times 10^2$	
Cs1	37	< 52	
Ra	226	< 111	
Th	226	< 87	
Th	332	< 216	

INDUSTRIAL ASPHALT CRYSTALLINE SILICA SOURCE TEST REPORT

REPORT OF FUGITIVE DUST MONITORING FOR CRYSTALLINE SILICA AT CALMAT FRESNO FACILITY

INTRODUCTION

From May 30 to June 27, 1990, personnel from Engineering-Science, Inc. (ES) conducted fugitive dust monitoring for crystalline silica at the following Calmat aggregate plants: Reliance, Fresno, Bakersfield, Pala, Mission Valley, Carol Canyon, Sloan Canyon, Durbin, Saticoy, Sun Valley, San Bernardino, and Palmdale. Grab samples of haul road, baghouse, and stock pile dust were collected in the months of May, June, July, August, and September by Calmat personnel at Fresno, Pala, Mission Valley, Carol Canyon, Sloan Canyon, Mojave and Colton. The samples were delivered to ES for analysis.

The monitoring was conducted at two locations at each plant. Sampling locations were usually the cone crusher and the screen that appear to generate the most dust. The ES testing technician was Mr. Rico Rivera. The grab samples were collected by Calmat personnel. The samples were taken at plant access roadsides, baghouse, and storage piles at the different facilities.

METHODOLOGY OF SAMPLING AND ANALYSIS AT FRESNO FACILITY

I. Sampling Procedure

A. Fugitive Dust Monitoring

High volume air samplers, with PM10 heads, properly located at the measurement site, drew a measured quantity of ambient air into a covered housing and through a tared polycarbonate filter during a 24 hour sampling period. Suspended particulates collected on the filter surface and were subsequently analyzed for crystalline silica content. The sample flow rate, collection times, and the increase in filter weight provided a measurement for the mass calculation.

B. Grab Samples

Grab samples of road haul dust were collected in jars from plant access roadsides. Grab samples of baghouse dust were collected in jars from the dryer/mixer baghouse at the industrial asphalt plant. No special equipment was used in grab sampling.

II. Analytical Procedure

Free Crystalline Silica Analysis (NIOSH Method 7500)

Sample Preparation -

The entire sample or an aliquot portion of the sample dust was suspended in 2-propanol and then agitated in an ultrasonic bath until all agglomerated particles were broken up. This suspension and all subsequent beaker washings were subjected to vacuum filtration through a 25mm silver membrane filter of $0.45\mu m$ pore size. This filter was mounted on an x-ray diffraction (XRD) sample holder for analysis.

Standards -

Standards were prepared by suspending 10.00mg and 50.00mg of the standard material, each into a 1 liter volume of 2-propanol. These suspensions were agitated in an ultrasonic bath for 20 minutes each. Aliquots were pipetted out and vacuum filtered onto silver membrane filters to produce working standard filters of varying sample sizes (e.g. $20\mu g$, $30\mu g$, $50\mu g$, $100\mu g$, $250\mu g$, etc.). These working standards were analyzed together with all samples and blanks.

Analysis -

Standards, samples, and blanks were qualitatively scanned from 10 to 80 degrees 2-theta by XRD. The areas under the peaks for each silica polymorph were measured over a long (e.g. 15 minute) scan time for each peak to allow low detection limits. The baseline measurements flanking each peak were taken in 1/2 of the peak scanning time.

Calculations -

Silica concentrations were calculated by comparing the intensity of the sample peak (corrected for background intensity and interferences) to the graph of standard intensities (also corrected for background intensity and interference). Concentration was calculated as the weight fraction of silica in total mass of particulate matter (particulates with diameter of less than $10\mu m$) and reported in mg/kg.

QUALITY ASSURANCE

I. Sample Custody

A specific Chain-of-Custody procedure was used for this project. The elements of this plan include:

- Sample identification
- Sample labels
- Documentation
- · Chain of Custody forms

The sequence of activities concerned with sample custody together with identification and tracking procedures are described below:

- 1. Filter preparation by laboratory, high volume sampler calibration and identification by tags and codes.
- 2. Filters issued to test team and master log filled out. Sample I.D. number stickers issued according to test identification code.
- 3. Filters recovered when a valid sample was obtained, accompanied by all field data sheets.
- 4. All samples returned to ES Pasadena laboratory with Chain-of-Custody form.
- 5. Samples examined at each transfer point for integrity and identity.

Upon completing the required analysis, the analyst returns the Chain-of-Custody form along with results to ES. All samples were accounted for by the ES Laboratory Supervisor and Project Manager. Each laboratory identifies samples in its own laboratory notebooks by the ES I.D. number as well as any internal identification. Notebooks were retained by each laboratory according to usual laboratory practices.

II. Calibration Procedures

The calibration procedures are specific to each analytical procedure. Standards are prepared from the highest grade reagents available, using procedures specified in the methods.

RESULTS

Analysis of all samples was performed by EMS Laboratories in South Pasadena. The results of the fugitive dust and grab sample testing are presented in Tables 1 and 2.

TABLE 1 SUMMARY OF RESULTS FOR ANALYSIS OF FUGITIVE DUST FOR CRYSTALLINE SILICA CONTENT AT CALMAT AGGREGATE PLANTS FROM MAY 30 TO JUNE 27, 1990

Source	Quartz	Crystalline Silica (Crystalobite	mg/kg) Tridymite
Mission Valley Secondary Crusher			
Final Screen	275,690	3,750*	3,750*
	256,540	2,620*	2,620*
Carol Canyon Secondary Crusher	145,270	3,380*	3,380*
Secondary Screen	169,740	4,060*	4,060*
Sloan Canyon Screen	64,900	6,500*	6,500*
Pala Primary Screen	24,100	6,020*	6,020*
Secondary Crus :	100,580	4,260*	4,260*
Fresno Secondary Crusher	56,620	4,480*	4,480*
Secondary Screen	255,730	4,280*	4,280 *
Palmdale Secondary Screen	260,400	4,260*	4,260*
Secondary Crusher	100,000	5,000*	5,000*
Bakersfield Secondary Crusher	191,720	23,000	360*
Primary Screen	184,770	16,600	830*
San Bernardino Secondary Crusher	186,000	18,770	430*
Primary Screen	171,000	20,446	2,970*
Sun Valley Secondary Crusher	249,240	3,340*	3,340*
Primary Screen	208,100	3,620*	3,620*
Reliance Secondary Crusher	164,180	7,460*	7,460*
Primary Screen	160,340	4,180*	4,180*
Saticoy Secondary Crusher	193,000	3,510*	3,510*
Primary Screen	168,420	5,260*	5,260*
Durbin Secondary Crusher	239,130	5,430*	5,430*
Secondary Screen	231,680	4,830*	4,830*

non-detected samples, value represents lower detection limit

TABLE 2 SUMMARY OF RESULTS FOR ANALYSIS OF ROAD, BAGHOUSE, AND STOCKPILE DUST

FOR CRYSTALLINE SILICA CONTENT AT CALMAT PLANTS

FROM MAY TO SEPTEMBER 1990

Source	Quartz	Crystalline Silica (1 Crystalobite	mg/kg) Tridymite
Mojave Iron Ore	11,000	2,000*	2,0() **
Clinker	3,000*	3,000*	3,000*
Limestone	2,000*	2,000*	2,000*
Silica	430,000	3,000*	3,000*
Bissell Clay	52,000	3,000*	3,000*
Baghouse Dust	49,000	3,000*	3,000*
Kiln Feed	50,000	3,000*	3,000*
Pacific Clay	34,000	3,000*	3,000*
Shale	150.000	3,000*	3,000*
Colton Iron Ore	86,000	3,000*	3,000*
Clinker	3,000*	3,000*	3,000*
Limestone	3,000*	3,000*	3,000*
Silica	920,000	2,000*	2,000*
Baghouse Dust	24,000	3,000*	3,000*
Kiln Feed	69,000	3,000*	3,000*
Catalyst Fines	3,000*	3,000*	3,000*
Shale	310,000	2,000*	2,000*
Mission Valley Solid	140,000	2,000*	2,000*
Haul Road Dust	120,000	3,000*	3,000*
Sloan Canyon Solid	78,000	2,000*	2,000*
Haul Road Dust	160,000	3,000*	3,000*
Carol Canyon Solid	190,000	2,000*	2,000*
Haul Road Dust	150,000	3,000*	3,000*
Pala Solid	90,000	2,000*	2,000*
Haul Road Dust	86,000	3,000*	3,000*
Fresno Industrial Asphalt Baghouse Dust	250,000	2,000*	2,000*
Fresno Haul Road Dust	160,000	2,000*	2,000*

^{*} non-detected samples, value represents lower detection limit